

REGATEC 2017

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Conference Proceedings

4th International Conference on Renewable Energy Gas Technology

22-23 May 2017, Pacengo (Verona), Italy

Editor: Jörgen Held













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22nd – 23rd of May 2017, Pacengo (Verona), Italy

Editor: Dr. Jörgen Held, Renewable Energy Technology International AB

Publisher

Renewable Energy Technology International AB Skarpskyttevägen 10 D SE-226 42 Lund SWEDEN info@renewtec.se

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Printed by Serviceförvaltningen/Tryckeriet Lunds kommun, Lund, SWEDEN, 2017 ISBN 978-91-981149-3-5

Direct injection mass spectrometry technique to monitor the removal of biogas trace compounds: biochar and ashes

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1. Introduction

Biogas from organic waste treatment could be used for the distributed energy production systems. SOFCs could be plausible due to high efficiency and reliability values. One of the main drawbacks concern the trace compounds impact. Waste derived materials as wood biochar and wood ashes could be adopted for the trace compounds removal. Their performance are monitored online with a PTR-ToF-MS instrument.

2. Material and methods

Organic fraction (OF) of municipal solid waste MSW was adopted as organic source to produce biogas with an anaerobic digester (AD) pilot plant located at FEM[1,2,3]. Table 1 summarizes the composition and physicochemical characteristics of the OFSMW batch used.

Table 1. Dry anaerobic digestion recipe.

	Mass	Water	Vol Sol	рН	
	(t)	(%)	(%)	in	off
Diges.	6.67	61.6	55.3	8.7	
OF + Wood	5.34	59.5	82	5.8	
Mix	12.0	58.8	59.8	7.8	8.3

Ashes from a forestry wood-chips boiler (3.3 MW, Viessman, Allendorf Germany) were tested in a glass reactor filter of 340 ml. Biochar from the pyrolysis of wood

was produced by Gruppo RM Impianti srl. (Italy), in a 200 kWe reactor at lower temperature 150 °C. The experimental setup is described in figure 1.

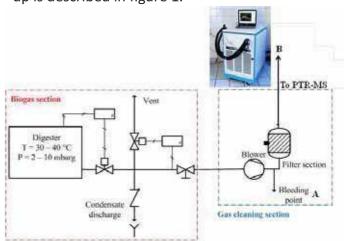


Figure 1. Biogas to gas cleaning section.

The pilot plant is equipped with pressure controller, automatic valves, a blower to increase the biogas pressure up to 80 mbar and a glass filter section. The PTR-MS adopted is described elsewhere [4,5].

3. Results

The study shows that char from the pyrolysis of wood waste is more promising for the trace compounds removal than using ashes as sorbent material. Ashes prove to be too dense, with an active surface area of 1 m²/g for the trace compounds removal compared to biochar. The active surface area and the microporous volume appear to be more important than the elemental composition for the removal performance. In the

following tables are reported the adsorption capacity for the most important trace compounds detected.

Table 2. Adsorption capacity - Biochar.

CH ₂ O	H ₂ S	C H 2 4 O	C ₂ H O ₆	CH₄S	C H 3 6 O	CH C 3 OOH
1.46	3.4	0.13	0.06	0.15	28.2	0.22
C₂H ₆ S	C₅ H ₈	C ₄ H ₈	C ₃ H ₃ S ₈	C H 6 6	C ₄ H ₈ O ₂	C ₄ H ₁₀
0.11	4.3	272. 01	0.76	0.71	0.81	13.35
C ₇ H ₈	C ₈	C H 8 1	C ₉ H	C H 10 1	C H 10 1	C ₆ H ₁₈ O ₃ Si ₃
433. 83	0.9	0.35	1.25	97.6	101. 6	13.15

Table 3. Adsorption capacity - Ash.

CH ₂	HS	C H 2 4 O	C ₂ H ₂ 6	CH₄S	C H 3 6	CH ₃ C OOH
0.0 01	0.0 14	0.00 7	0.01 5	0.00	0.03	0.001
C₂H ₆ S	C ₅ H	C ₄ H ₈ O	C H 3 8	C ₆ H	C ₄ H ₈ O ₂	C ₄ H ₁₀ S
0.0 01	0.0	0.20 7	0.00	0.00	0	0.045
C7 H8	C ₈ H 8	C H 8 1	C H 9 1	C 10 H	C ₁₀ H ₁₆	C ₆ H ₁₈ O ₃ Si ₃
0.3 11	0.0 01	0.00	0.00	0.23	0.24	0

The adsorption capacity, for example of H_2S using biochar shows a value around 3.39 mg/g and 0.014 mg/g using ashes.

4. Conclusions

The biogas coming from the AD of OFMSW was sent to a gas cleaning section was filled with waste derived materials: biochar and ashes. The best performance is achieved considering biochar as sorbent material. This is due to the structure of the sorbent more than the elemental composition. In fact, elements able to remove trace compounds are contained in larger way in ash sample. The active surface area and microporous volume play a crucial role for the trace compounds removal. Future works are related to the activation procedures for the char in order to compare it with commercial sorbents.

5. References

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