LC-MS METABOLOMICS SHOWS A SMART WAY TO REDUCE SULFITES IN WINE

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Introduction

The impact of minute amounts of headspace oxygen on the postbottling development of wine is generally considered to be very important, since oxygen, packaging and storage conditions can either damage or improve the quality of wine, in terms of its characteristics.

Experimental Design

The sample set included 12 white wines from 6 varieties. 10 samples of each wine were bottled using the standard industrial process, with inert headspace and variable exposure to oxygen (Low O_2), along with a further 10 bottles produced using the same line, but without inert gas and with extra headspace (High O_2). The wines were analyzed after two months of storage.

High O₂ Variety Code Low O₂ Wine Pinot gris x 5 x 10 PNT x 10 Grillo GRL x 3 x 10 x 10 Chardonnay CHR x 10 x 10 x 1 Muller Thurgau x 10 MLR x 1 x 10 Traminer x 10 TRM x 1 x 10 Inzolia INZ x 10 x 1 x 10

Total 240 bottles

Bottles bottled

Results I

concentrations of total The mean package oxygen (TPO) in each wine under the two bottling conditions (respectively, High vs Low O_2) at the time of bottling and after two months of storage.



NomaSense[™] oxygen analyzer





Low O₂

High O₂

(N)



multivariate statistical





Results III

New reactions discovered: The main reactions driven by O_2 were involving Sulfur. The antioxidant SO₂, added to protect wine from unwanted reactions, takes part in various reactions, several of which were unknown in wine to date and would appear to be of practical significance.

Possible additive interaction effects, between SO₂ and glutathione, should be revaluated, since these could also turn out to be antagonistic, and their coaddition/co-presence could provide less effective protection.

Wines containing large amounts of indoles might require the addition of larger amounts of SO₂ or/and should be bottled under low oxygen conditions.





The co-presence of glutathione (GSH) and SO₂ in wine has as a result the formation of S-sulfonated glutathione (GSSO₃H). The mechanism of the reaction requires first the oxidation of GSH to GSSG (glutathione disulfide), which reacts very fast to produce GSSO₃H. This is the cause why GSSG is not detectable in wine.

For further information read: Arapitsas, P.; Ugliano, M.; Perenzoni, D.; Angeli, A.; Pangrazzi, P.; Mattivi, F. Wine metabolomics reveals new sulfonated products in bottled whitewines, promoted by small amounts of oxygen, Journal of Chromatography A, vol. 1429, pp. 155-165, 2016.