

Annual Meeting 2024 April 25 & 26

Amsterdam, the Netherlands



25/04 -13:15 - 13:30

How to guarantee authenticity and traceability of agri-food and supplements products thanks to the application of isotopic analysis of bioelements

¹ Perini Matteo, ¹ Pianezze Silvia

¹ Fondazione Edmund Mach, Via Mach 2, 38098 San Michele all'Adige (TN), Italy

Stable isotope ratio analysis of bio-elements (hydrogen, carbon, nitrogen, oxygen and sulphur) has been used since the 1990s to check food authenticity and traceability of a wide variety of food commodities (Rossmann, 2001). In the last few years, examples of applications also in the pharmaceutical and cosmetic field have been reported (Pellati et al., 2013; Perini et al., 2017, 2021; Perini, Paolini, et al., 2019; Perini, Pianezze, et al., 2019). The use of stable isotope analysis for products authentication purposes is possible thanks to isotopic fractionation occurring in several processes and reactions (biological, biochemical, physical, chemical etc.) which generates unique isotopic signatures. For this reason, the application of this technique on the bulk samples as well as on specific components (e.g. aroma compounds) can be used to detect the origin of an ingredient (synthetic or natural), the substitution of one ingredient for another, as well as the geographical and/or botanical origin of the products.

The widespread and well-known technique based on the coupling between elemental analyzer and mass spectrometer (EA-IRMS) is now flanked by liquid chromatography (LC-IRMS) and gas chromatography (GC-IRMS). Today it is therefore possible to analyze not only the bulk of the matrices but also their individual components.

The δ^{13} C and δ^{2} H values of vanillin can determine whether the product is natural (deriving from the expensive CAM plant Vanilla), biotechnologically derived or synthetic (Perini, Pianezze, et al., 2019). Moreover, the δ^{13} C values of specific components of Rosa damascene mill., one of the most expensive essential oils in the market world, can indicate the fraudulent addition of cheaper oil from a C4 plant (e.g. Cymbopogon martinii, palmarosa) (Pellati et al., 2013).

In pharmaceutical and cosmetic formulations, δ^{13} C analysis is a suitable tool to discriminate between squalene and squalane from shark (illegal) and from olive oil (expensive) (Camin et al., 2010) as well as between monacolin K (contained in the fermented dietary supplement red yeast rice) and the commercially marketed statin, lovastatin (Perini et al., 2017).The L-theanine extracted from Camellia Sinensis is easily distinguishable from that obtained biosynthetically (Perini et al., 2021).

It is possible to combine different isotopic signatures to guarantee the natural origin of curcumin, caffeine (Ding et al., 2019), tartaric acid and its derivatives.

These examples demonstrate that the isotopic fingerprint represent an effective tool for the authenticity assessment of economically important pharmaceutical, cosmetic and supplement products.

References

Camin, F., Bontempo, L., Ziller, L., Piangiolino, C., & Morchio, G. (2010). Stable isotope ratios of carbon and hydrogen to distinguish olive oil from shark squalene-squalane. Rapid Communications in Mass Spectrometry: RCM, 24(12), 1810–1816. Ding, B., Zeng, G., Wang, Z., Xie, J., Wang, L., & Chen, W. (2019). Authenticity determination of tea drinks in the Chinese market by liquid chromatography coupled to isotope ratio mass spectrometry. Microchemical Journal, Devoted to the Application of Microtechniques in All Branches of Science, 144, 139–143.

Pellati, F., Orlandini, G., van Leeuwen, K. A., Anesin, G., Bertelli, D., Paolini, M., Benvenuti, S., & Camin, F. (2013). Gas chromatography combined with mass spectrometry, flame ionization detection and elemental analyzer/isotope ratio mass spectrometry for characterizing and detecting the authenticity of commercial essential oils of Rosa damascena Mill. Rapid Communications in Mass Spectrometry: RCM, 27(5), 591–602.

Perini, M., Carbone, G., & Camin, F. (2017). Stable isotope ratio analysis for authentication of red yeast rice. In Talanta (Vol. 174, pp. 228–233). https://doi.org/10.1016/j.talanta.2017.05.057

Perini, M., Paolini, M., Pace, R., & Camin, F. (2019). The use of stable isotope ratio analysis to characterise saw palmetto (Serenoa Repens) extract. Food Chemistry, 274, 26–34.

Perini, M., Pianezze, S., Strojnik, L., & Camin, F. (2019). C and H stable isotope ratio analysis using solid-phase microextraction and gas chromatography-isotope ratio mass spectrometry for vanillin authentication. Journal of Chromatography. A, 1595, 168–173.

Perini, M., Pianezze, S., Ziller, L., & Camin, F. (2021). Characterization of L-theanine in tea extracts and synthetic products using Stable Isotope Ratio Analysis. In Journal of Food and Drug Analysis (Vol. 29, Issue 2, pp. 312–319). https://doi.org/10.38212/2224-6614.3349

Rossmann, A. (2001). DETERMINATION OF STABLE ISOTOPE RATIOS IN FOOD ANALYSIS. In Food Reviews International (Vol. 17, Issue 3, pp. 347–381). https://doi.org/10.1081/fri-100104704